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Key indicators

Single-crystal X-ray study
 T = 100 K
 Mean $\sigma(C-C)$ = 0.002 Å
 R factor = 0.035
 wR factor = 0.112
 Data-to-parameter ratio = 27.6

For details of how these key indicators were
 automatically derived from the article, see
<http://journals.iucr.org/e>.

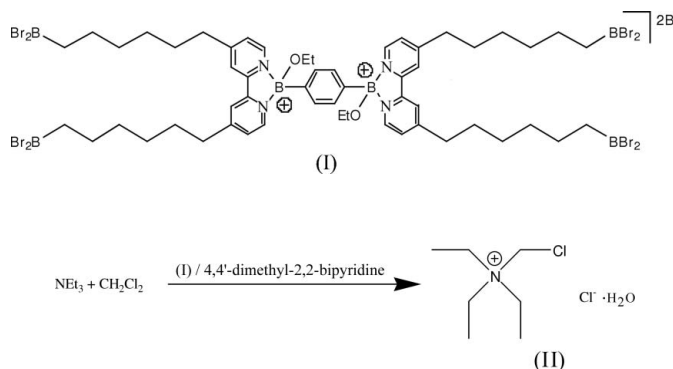
**(Chloromethyl)triethylammonium chloride
 monohydrate**

The title compound, $C_7H_{17}ClN^+ \cdot Cl^- \cdot H_2O$, is composed of discrete (chloromethyl)triethylammonium cations, Cl^- anions and water molecules, which are held together by $O-H \cdots Cl$ hydrogen bonds.

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Comment

Treatment of a solution of the diboronium, (I) (Haberecht, 2006), and 4,4'-dimethyl-2,2'-bipyridine in CH_2Cl_2 with NEt_3 led to the formation of the title compound, (II), $(Et_3NCH_2Cl)^+(Cl)^- \cdot H_2O$.



Compound (II) consists of discrete (chloromethyl)triethylammonium cations, Cl^- anions and water molecules. Geometric parameters adopt normal values (Cambridge Structural Database, Version 5.27, updated May 2006; *MOGUL* Version 1.1; Allen, 2002). Two water molecules and two Cl^- anions form a hydrogen-bonded eight-membered centrosymmetric ring (Fig. 2). In addition to these classical hydrogen bonds, there are also intramolecular $C-H \cdots Cl$ contacts, as well as intermolecular $C-H \cdots O$, $O-H \cdots Cl$ and $C-H \cdots Cl$ contacts (Table 1).

Experimental

To a solution of diboronium dibromide, (I) (1 mmol), and 4,4'-dimethyl-2,2'-bipyridine (0.7 g, 4 mmol) in CH_2Cl_2 (40 ml), NEt_3 (1.5 ml) was added. $(Et_3NCH_2Cl)^+(Cl)^- \cdot H_2O$ was obtained as a side-product. X-ray quality crystals of the title compound were grown from a solution in CH_2Cl_2 at ambient temperature.

Crystal data

 $C_7H_{17}ClN^+ \cdot Cl^- \cdot H_2O$ $M_r = 204.13$ Monoclinic, $P2_1/n$ $a = 11.1642(8) \text{ \AA}$ $b = 7.9798(4) \text{ \AA}$ $c = 12.3239(9) \text{ \AA}$ $\beta = 106.087(6)^\circ$ $V = 1054.92(12) \text{ \AA}^3$ $Z = 4$ $D_x = 1.285 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation $\mu = 0.57 \text{ mm}^{-1}$ $T = 100(2) \text{ K}$

Block, colourless

 $0.50 \times 0.40 \times 0.30 \text{ mm}$

Data collection

Stoe IPDS-II two-circle diffractometer

 ω scans

Absorption correction: multi-scan (MULABS; Spek, 2003; Blessing, 1995)

 $T_{\min} = 0.764$, $T_{\max} = 0.848$

26921 measured reflections

2977 independent reflections

2664 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.032$ $\theta_{\text{max}} = 29.7^\circ$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.035$ $wR(F^2) = 0.112$ $S = 1.06$

2977 reflections

108 parameters

H atoms treated by a mixture of independent and constrained refinement

 $w = 1/[\sigma^2(F_o^2) + (0.0699P)^2 + 0.5992P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} < 0.001$ $\Delta\rho_{\text{max}} = 0.81 \text{ e \AA}^{-3}$ $\Delta\rho_{\text{min}} = -0.68 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$O1-H1D \cdots Cl1$	0.83 (1)	2.49 (1)	3.3051 (13)	170 (3)
$O1-H1C \cdots Cl1^i$	0.83 (1)	2.34 (1)	3.1660 (13)	172 (3)
$C7-H7A \cdots O1$	0.99	2.36	3.3131 (19)	161
$C7-H7B \cdots O1^{ii}$	0.99	2.57	3.429 (2)	145
$C1-H1A \cdots Cl1^{iii}$	0.99	2.75	3.6809 (15)	156
$C1-H1B \cdots Cl1^{iv}$	0.99	2.76	3.7471 (15)	176
$C3-H3A \cdots Cl17$	0.99	2.72	3.1677 (15)	108
$C5-H5B \cdots Cl17$	0.99	2.75	3.0704 (18)	100

Symmetry codes: (i) $-x+1, -y, -z+1$; (ii) $-x+\frac{1}{2}, y+\frac{1}{2}, -z+\frac{1}{2}$; (iii) $x-\frac{1}{2}, -y+\frac{1}{2}, z-\frac{1}{2}$; (iv) $-x+\frac{3}{2}, y+\frac{1}{2}, -z+\frac{1}{2}$.

H atoms bonded to C atoms were located in a difference map but were subsequently refined with fixed individual isotropic displacement parameters [$U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$, or $1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$] using a riding model, with $C-H = 0.98$ and 0.99 \AA for methyl and methylene groups, respectively. The water H atoms were refined with a distance restraint of $O-H = 0.84(1) \text{ \AA}$.

Data collection: *X-AREA* (Stoe & Cie, 2001); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

References

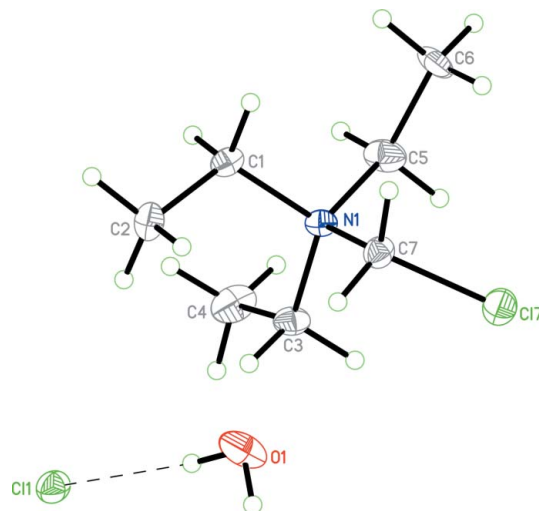
Allen, F. H. (2002). *Acta Cryst.* B58, 380–388.

Figure 1

A perspective view of the asymmetric unit of (II), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. The dashed line indicates a hydrogen bond.

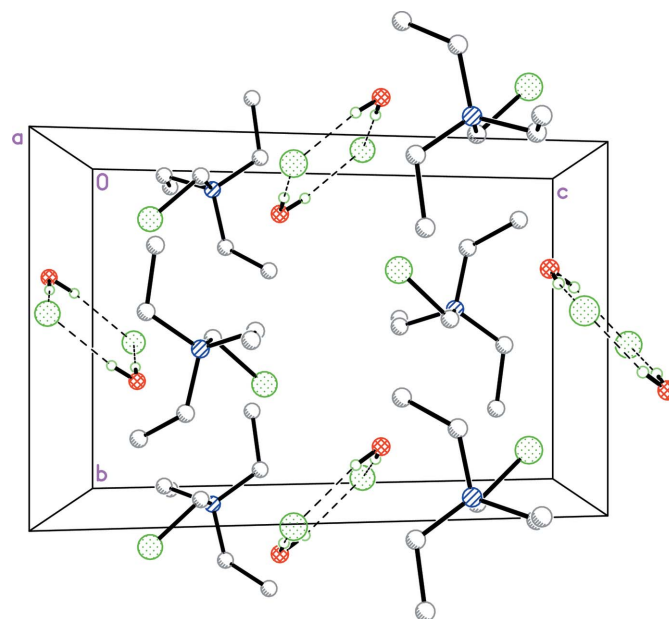


Figure 2

A packing diagram for (II), viewed along the a axis. H atoms bonded to C atoms have been omitted for clarity. Dashed lines indicate hydrogen bonds.

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